## THE MERCK INDEX

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### Ethyl a-Bromopropionate

3870

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H.SO. Kamm, Marvel, Org. Syn. coli. vol. I, 29 (1941). Physical properties: Mumford, Phillips, J. Chem. Soc. 1950, Physical properties: Mumford, Phillips, J. Chem. Soc. 1950, 75. Toxicity data: E. H. Vernot et al., Toxicol. Appl. Pharmacol. 42, 417 (1977).

Coloriess, flammable, volatile liq; ethereal odor; burning Colorless, flammable, volatile liq; ethereal odor; burning taste; becomes yellowish on exposure to air and light. Vapor harmful. d20 1.4612; d2 1.4515. bp 38.2°. mp -119°. n20 1.4242. Soly in water (g/100 g) at 0°: 1.067; 10°: 0.965; 20°: 0.914; 30°: 0.896; miscible with alcohol, ether, chloroform and with other organic solvents. Explosive limits (% by vol in sir), lower 6.75, upper 11.25. Auto-ignition temp 952°F (511°C). LC<sub>50</sub> rats, mice (ppm): 27000, 16200 (Vernot).

Caution: Potential symptoms of overexposure are irrita-

Caution: Potential symptoms of overexposure are irrita-tion of eyes, respiratory system and skin; central nervous system depression; pulmonary edema; liver and kidney dis-ease; cardiac arrhythmias; cardiac arrest. See NIOSH Pocket Guide to Chemical Hazards (DHHS/NIOSH 90-117, 1900 - 105

1990) p 106.

USE: Ethylating agent in organic synthesis: as refrigerent.

Pornerly used as a topical and inhalation anesthetic.

3820. Ethyl a-Bromopropionate. 2-Bromopropanoic acid ethyl ester. C<sub>4</sub>H<sub>2</sub>BrO<sub>3</sub>; mol wt 181.03. C 33.17%, H 5.01%, Br 44.14%, O 17.68%. CH<sub>3</sub>CHBrCCOC<sub>2</sub>H<sub>3</sub>. Liquid; sharp, pungent odor; becomes yellow on exposure to light. d<sup>20</sup><sub>10</sub> 1.447. bp 159-160'; also stated as 160-165'. n<sup>20</sup><sub>10</sub> 1.4469. Insol in water: miscible with alcohol, ether. Protect from light.

3821. Ethyl terr-Butyl Ether. 2-Ethoxy-2-methylpropane; terr-butyl athyl ether; ethyl terr-butyl oxide: 1,1-dimethylethyl ether; ethyl 1,1-dimethylethyl ether; ETBE. C.H.,O; mol wt 102.18. C 70.53%, H 13.81%, O 15.66%. (CH.)-C(OCH,CH.). Prepn I. U. Ncf. Ann. 309, 126 (1899). Synthesis: J. F. Norris, G. W. Rigby, J. Ann. Chem. Soc. 54, 2088 (1932). Physical properties: T. W. Evons, K. R. Edlund, Ind. Eng. Chem. 28, 1186 (1936). Thermal decomposition: N. I. Daly, C. Wentrup, Aust. J. Chem. 21, 1535 (1968). Brief review focusing on use as gasoline additive: M. Iborra et al. Chemtech. 18, 120 (1988). bp 69-71°. fp -94.0°. Also reported as bn. 73.1° (Norris

(1988).

bp 69-71°. fp —94.0°. Also reported as bp 73.1° (Norris, bp 69-71°. fp —94.0°. Also reported as bp 72.8°. Rigby). d\(^3\) 0.7364. n\(^3\) 1.3728. Also reported as bp 72.8°. (Evans. Ediund). d\(^3\) 0.7456; d\(^3\) 0.7404; d\(^3\) 0.7353, d\(^3\) 0.7300. n\(^3\) 1.3760. Vapor pressure at 25°: 130 mm Hg. Heat vaporization: 74.3 cal/g. Specific heat (liquid) at 25°: 0.51 cal/g/°C. Surface tension at 24°: 19.8 dynes/cm. Soly in water (20°): 1.2 g/100 g soln. Soly of water in compound (20°): 0.5 g/100 g soln.

USS: Gasoline additive. USB: Gasoline additive.

3822. Ethyl Butyrate. Butanoic acid ethyl ester; butyric acid ethyl ester; ethyl n-butyrate. C.H.;O; mol wt 116.16. C 62.04%, H 10.41%, O 27.55%. CH.;CH.;CH.;CO-OC,H. Toxicity data: P. M. Jenner et al. Food Cosmet. Toxicol. 2, 327 (1964).

Coloriest lie: pineapple oder. 429 0.829. bp. 120.131

Colorless liq: pineapple odor. dip 0.879. bp 120-121'. mp -93'. np 1.400. Flash pt, closed cup: 78°F (25°C); open cup: 85°F (29°C). Sol in about 150 parts water; misc with alcohol, other. LD<sub>80</sub> orally in rats: 13,050 mg/kg (Senger)

(Senacr).

USE: Manuf artificial rum; perfumery; the alcoholic solu constitutes the an-called \*pineapple oil\*.

3823. Ethyl Caprate. Decanoic acid only! ester; ethyl decanoate. C<sub>1</sub>H<sub>3</sub>O<sub>2</sub>; mol wt 200.32. C 71.95%, H 12.08%, O 15.97%. CH<sub>3</sub>CH<sub>2</sub><sub>1</sub>COOC<sub>2</sub>H<sub>5</sub>. Colorless liq. d<sup>20</sup> 0.862. bp 243-245°. Insoluble in water; missible with alcohol, chloroform, ether.

USE: Manuf wine bouquets, coguac essence

3824. Ethyl Caproate. Heranoic acid sthyl ester, ethyl heranoatc. C<sub>6</sub>H<sub>2</sub>O<sub>7</sub> mol wt 144.21. C 66.63%, H 11.18%, O 22.19%. CH<sub>3</sub>(CH<sub>2</sub>)<sub>4</sub>COOC<sub>2</sub>H<sub>5</sub>.

Colorless to yellowish liquid; pleasant odor. d<sup>20</sup> 0.871 bp 166-167. Insol in water, miscible with alcohol, other USE: Manuf artificial fruit flavors.

3825. Ethyl Caprylate. Octanoic acid ethyl ester; calculoctanoate; ethyl octylate: C<sub>10</sub>H<sub>20</sub>O<sub>1</sub>; mol wt 172.27. 69.72%, H 11.70%, O 18.58%. CH<sub>3</sub>(CH<sub>2</sub>)<sub>C</sub>COOC<sub>2</sub>H<sub>3</sub>. Colorless, clear, very mobile liquid; pleasant, pincapple odor, dt<sup>1</sup> 0.878. bp 207-209°, Insol in water; mise with all ether. LD<sub>50</sub> orally in rats: 25.960 mg/kg, P. M. Jenner of al. Food Cosmer. Taxicol. 2, 327 (1964).

USE: Manuf fruit ethers; consuit of enanthic, coccie, and cognae ethers.

cognac ethers.

3826. Ethyl β-Carbolina-3-carboxylate. 9H-Pyrido(). 4-b]indoie-3-carboxylic acid othyl ester; ethyl norharmancur-boxylate; β-CCE. C<sub>14</sub>H<sub>14</sub>N<sub>2</sub>O<sub>2</sub> mol wt 240.26. C 69.9°:. H 5.03%, N 11.66%, O 13.32%. Deriv of β-carboline that is a potent displacer of H-diazepam from brain benzodinzea potent displacer of M-diazepam from brain benzodiazepine receptors. Isoln from human urine and brain and hinding site study: C. Bracstrup et al., Proc. Nat. Acad. Sci. USA 77, 2288 (1980). Initially thought to be an endogenous ligand for benzodiazepine receptors in mammalian CNS, it is now believed to be formed during isoln and extraction is now believed to be formed during isoln and extraction is now believed to be formed during isoln and extraction is now believed to be formed during isoln and extraction is now believed to be formed during isoln and extraction is now believed to be formed during isoln and extraction is now believed to be formed during isoln and extraction is now believed to be formed during isoln and extraction is now believed. (Raven Press, New York, 1980) pp 129-138; M. Nielson et al., J. Neurochem. 36, 276 (1981). Synthesis and psychotropic activity: Japan, Kokkal 81 43283 (to Schering AG), G.A. 95, 115508a (1981); U. Eder et al., Eur. pat. Appl. 30,254 (1981 to A/S Perrosan; Schering AG). β-CCE has been shown to lower seizure threshold and to reverse the sedative effect of flurazepam, q.V. P. J. Cowen et al., Nature 290, 54 (1981). Neurochemical and pharmacological actions of β-CCE and other β-carbolines; M. Cain et al., J. Med. Chem. 25, 1081 (1982). Anxiogenic and convulsant properties: L. Prado de Carvalho et al., Nature 301, 64 (1983).

mp 229-233°. uv max (pH 7): 215, 242, 279 nm. mp 229-233°. uv max (pH 7): 215, 242. 279 nm.
3-Hydroxymethyl-β-carboline, C<sub>12</sub>H<sub>10</sub>N<sub>1</sub>O, 9H-pyrido[3,4-b]Indole-3-methanol, 3-HMC. Prepn: F. Hamnguchi, S. Ohki, Heterocycles 8, 383 (1977); M. Cain et al., loc. cit.
Antagonism of anticonvulsant and anxiolytic actions of diazepam: P. Skolnick et al., Eur. J. Pharmacol. 68, 381 (1980). Crystala, mp 225-228°.

USE: As tools for studying benzodiazepine receptors.

USE: As tools for studying benzodiazepine receptors.

3827. Ethyl Carbonate. Carbonle acid diethyl estar; diethyl carbonate; Eufin. C<sub>3</sub>H<sub>10</sub>O<sub>2</sub>; mol wt 118.13. C 50.84%, H 8.53%, O 40.63%. (C<sub>2</sub>H<sub>3</sub>O)<sub>2</sub>C=O. Prepn. Paloman et al., Ber. 72, 313 (1939). Manuf: Mador, Blackham, U.S. pat. 3,114,762 (1963 to Natl. Distillers & Chem.). Liquid, bp 126°. Pleasant ethereal odor, mp - 43°. Flash pt, closed cup: 777F (25°C). d<sup>26</sup> 0.9764. n<sup>26</sup> 1.3843. Practically insol in water, miscible with alcohol, ether. USE: Solvent for nitroccilulose; manuf radio tubes; fixing rare earths to cathode elements.

rare carths to cathode elements.

3828. Ethyl Cellulose. Cellulose ethyl ether, Ethocci. Prepd from wood pulp or chemical cotton by treatment with alkali and ethylation of the alkali cellulose with ethyl chloride. Review and bibliography: E. Ort. Cellulose and Cellulose Derivatives (New York, 2nd ed., 1955).

White granules. Soly is dependent upon the degree of substitution. Commercial ethyl cellulose has an ethoxy content of 43-50%. A 47% product softens at 140° and is sol in ethyl acetate, ethylene dichloride, benzene, toluene, xylene,

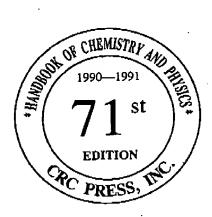
tent of 43-50%. A 47% product softens at 14U and is sol in ethyl acetate, ethylene dichloride, benzene, toluene, xylene, butyl acetate, acetone, methanol, chanol, butanol, carbon butyl acetate, acetone, methanol, ethanol, butanol, carbon but acetate, acetone, methanol, ethanol, butanol, carbon but acetate, acetone, acetate, ethyl cellulose formulations usually include an antioxidant such as hydroquinone monobenzyl ether, 4-hexylpyrocatechol, or diphenylamine,

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PHYSICAL CONSTANTS OF ORGANIC COMPOUNDS (Continued)										
No.	Name, Synonyms, and Permula	Mol. wt,	Color, crystalline form, specific rotation and 1 (log r)	ь.g.	<sup>™</sup> 'S.	Density	ft <sub>D</sub>	Solubility	Rof.	
830	Cellulose, hexanitrate of Gun contant.	(594.27)	whamor		(60-70 (ign)	1.66		ph NO,		
831	(C.,H.,N,O.), Cellulosa, pentanitrate	(549, 28)	wh amor		1.66			esh-al		
	(C,1H,1N,O.1). Cellulose, retranitrate or in Collodion	(504.28)	wh amot	<b> , . ,</b> ,		1.66		eth-al	• • • • • • •	
	(Ci,Hi,N,Oil.	(288.25)	yeshil [ø]n					48		
	(C,,H,,O,).		-22.5 (chl)		240-55		<b>,,</b> .	cth		
B34	Cellulose, triethytether or Ethylcellulose	(246.30)	( a) *** a + 26.1 (bz)							
875	Cellulose, trinitrate or in Collodion	(4\$9.28)		<b></b> .		1,66		acc, &8		
	(Ç <sub>3</sub> ,H <sub>3</sub> N <sub>2</sub> O <sub>3</sub> ). Lepnaruninine		λευωοι b.e.	,,	145-5	, , , , , , , , , ,		al, esh, acc. bz	C49, 1745	
	C,,H,,N,O,		+ 277 (chl)	207" '	61 .			al, eth	BI', 143	
837	Cerane or Isohexacosane	366.71	pl (eth), sc (w)	207"					B8', 573	
<b>8</b> 28	Cerulignone	304.30	ng fo		18.2	0.7733***	1,4345		131-, 537	
839	Ceune or Hexadecone	226.45	If (ace)	287, J49'"	1	0.781144	1.4412 <sup>m</sup>	l	BI*, 927	
	Cetene or 1-Hexaderene		st.	284.4, 155"	4.1	0.B176****	1.4283**	eth, ace, bz, chl	B1*, 1876	
	Cetyl alcohol or I-Hexadecanol		(I (AcDEI)	344, 190"	50	I	1.449634	al, eth, ace, b2,	1	
842	CH.(CH.), CH.OH  Cetylamine or 1-Amino hexadecone	241.46	tt	323.5. 144'	46.8	0.8129***		chi	B6*, 555	
1843	CH <sub>1</sub> (CH <sub>1</sub> ), CH <sub>1</sub> NH, Cetyl Phonyl Ether or Hexadecyl phonyl ether,	318.54	If (al)	200'	41.8	0.84344	1.4556**		i	
	Civil sulfate		,		. 66.2			- "	B1*, 1879	
114	(C,H,,Q),SO,	509.64	nd (MeOH- eth), [ a ]		- 246-B		•	• • • • • • • • • • • • • • • • • • • •	B21*, 681.	
	C,, H,, NO,		-47,5 (al)	١,	1			1.		
12-41	Chalcone dibromide (threu)	168,07	(la) bn		. 122-1			. al	B7*, 2155	
	Callacte dibromide (crythro)	. 365.07	prored (#I)		. 159-60				874, 2154	
	C.H.CHBrCHBrCOC.H. Chalcone-(trans) of Benzalacetophenone		pa ye lf.	345-84. 208"	(i)59 (ii)57	1.0712****	,	eth, bz, chl	B7', 2380	
<b>1</b> (340)	C'H'COCH-CHC'H'	1	pr. nd (peth)		(111)49	1	,,	al	B7°, 441	
184	Chalcone, 4,4-dimethyl ,	1	cr (MeOH)		210-1			bz	B7*, 2407	
	Chalcone, 3.J'-dinitrotro	1	pa ye nd (pa)					al, cih, bz, chi	BB*, 1450	
485	. Chalegoe 2-methoxy of 2-Anssylident acetaphenone	238.29	yesh aid (peth or	.,	. 64-5		1			
	1 Chalcone, 1-methococ, H.	238.29	eth-tig)	24711	65	ļ ,		, al, eth. acc. Dz	B8*, 146	
445	1 Chalcone, 1-methoxy (3-CH,0C,H,)CH=CHCOC,H,		(McOH)					. al, eth, chi. aa	98°, 146	
185	3 Chakone, 4-methoay	. 238.29	ye no (al)	187-81	79		1	1.	9194, 18	
	Chalcone 1.4 methylene digry or Piperonylidine aucto-	252.27	ye ad (al)		. 128			.   a], au	1	
	phenone HICH-CHCOC.H.	253.26	pa bind		. 125			, al, eth, as	B7', 239	
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	6 Chalcone, 2'-nitro	253.26		1		1		. al, bz, chl. 89	B7', 240	
1A4	7 Chalcone, 3-nitro		ps)	Ţ	1		1	1	240 , יקפ	
	# Chalcone, 4-chtro H-O,NC,N,ICH-CHCOC,H.	, 253.26	payend (al). pl(bz)		1	1	1	1		

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CH (CH ) CO.C.H. FW 172.27	5ġ -	9.10 19.90
11,232.1 Ethyl captylate, 99 + 76 1104 00,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Metals mp 48 to 47° bp 208-208° nB 1.4170 d 0,878 Fp 167 °F(75 °C) Bell, 2,948 Fp 167 °F(75 °C) Bell, 2,948 Fp 167 °F(75 °C) Bell, 2,948 Fp 167 °F(75 °C)	00g 00g	08.69
Ethyl carbamate, 586 U2857, Urettina PF86 Ethyl carbazata, 97% [4114-31-2] H <sub>3</sub> NNHCO,C <sub>2</sub> H <sub>3</sub> , FW 104.11 mp 44-47° Ethyl carbazata, 97% [4114-31-2] H <sub>3</sub> NNHCO,C <sub>2</sub> H <sub>3</sub> , FW 104.11 mp 44-47° bp 108-110°/22mm Fp 187°F(88°C) 86/I. 3,98 Flaser 1,350 FT-NMR 1(1),1286C bp 108-110°/22mm Fp 187°F(88°C) 86/I. 3,98 Flaser 1,350 FT-NMR 1(1),1286C	25g	19,10
E1,660-0 9-Ethylcarbazole, 98% [86-28-2] FW 195.27 mp 68-70° Boil. 20,436 E1,660-0 9-Ethylcarbazole, 98% [86-28-2] FW 195.27 mp 68-70° Boil. 20,436	5g 100g 500g 25g	17.10 52.00 173,20 94.20
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Ethyl β-cerboline-3-carboxylate, 355 and 1997 and 511		18.40
Ethyl carbonate, see D#155-1, Dietryl carbonate, see D#155-1,	5g 250g 500g	60.70 109.40
20,064-6 Ethyl cellulose [9004-57-3] Softening point 155°C. Viscosity (5% solution in	5g 250g 500g	23.00 76.90 126.60
20,068-9 Ethyl cellulose [8004-57-3] Softening point 155°C. Viscosity (5% solution in	5g 250g 5 <b>9</b> 0g	17,10 56,30 93,90 17,10
20,069-7 Ethyl cellulose [9004-57-3]	5g 250g 500g	56.30 93.90 17.10
43,983-7 Ethyl cellulose (9004-67-3)	5g 250g 500g	56.30 93.90 19.50
24,749-9 Ethoryl content 48%. Softening point 155°C. Viscosity (5% solution in	5g 100g 500g 5g	29,10 107,30 17,10
20,085-4 Ethyl cellulose [9004-57-3]	250g 500g	56.30 93.90 23.00
# Granules, Ethoxyl contain 300 cps  18,102-1 Ethyl cellulose [9004-57-3].  * Powder, Ethoxyl contain 49%. Softening point 182°C, Viscosity (5% solution in powder, Ethoxyl contain 49%.	250g . 5g	23.00
20,066-2 Ethyl cellulose [9004-57-3]		75.90 126.60
Ethyl chloride, see Chloroethane  Ethyl chloride, see Chloroethane  [26226-72-2] CICH, CONHC, H. CO, C, H.	59 259 19	25.20 84.40 12.60
44,543-8 Ethyl 2-(2-chloroacetamino).4-titip20188000ths, 5-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1	10g 100g	69.50 27 <b>.3</b> 0
* No. 1.420 FB 13.A.1 Salety 2,15888 R&S 1(1),735A HIELS# AFOTTOBER FT-IR 1(1),650D S/113.A.1 Salety 2,15888 R&S 1(1),735A HIELS# AFOTTOBER HIGHLY TOXIC LACHRYMATOR		
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